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## EE 527 MICROFABRICATION

Lecture 22  
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### WET ETCHING OF ALUMINUM (AL)

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- $\text{HCl} : \text{H}_2\text{O} @ 1:4$
- $\text{H}_3\text{PO}_4 : \text{HNO}_3 : \text{H}_2\text{O} @ 74:7:19$  gives a very rapid etch rate
- Almost any of the strong acids with an oxidizer will etch aluminum:
  - $\text{H}_2\text{SO}_4$
  - $\text{HNO}_3$
  - $\text{HCl}$
  - $\text{H}_3\text{PO}_4$
- The trick is usually diluting the acid / oxidizer mixture down to a point to where it can be controllable.
- If the aluminum has a thick skin of  $\text{Al}_2\text{O}_3$ , this may first have to be removed using an HF-based etchant.



## WET ETCHING OF GOLD (AU)

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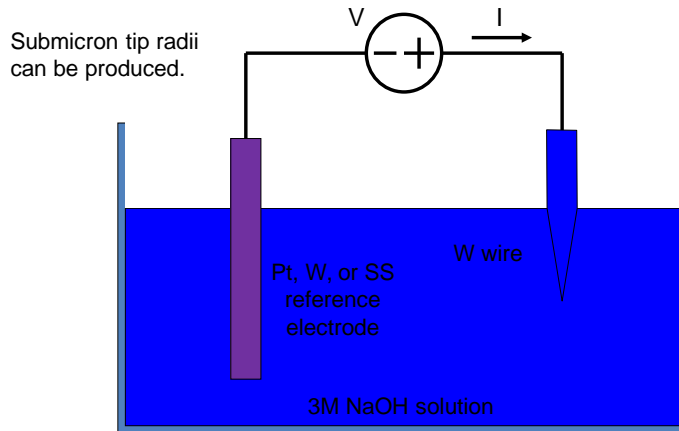
- HCl : HNO<sub>3</sub> @ 3:1 (Aqua regia)
- Gold Etch (TFA):
  - KI : I<sub>2</sub> : H<sub>2</sub>O @ 1.3 g : 65 g : 100 mL @ room temperature (2-0-0)
  - Etch solution is nearly black and impossible to see wafer once submerged.
  - Also etches most compound semiconductors @ ~1.0-2.0 μm/min

## HYDROXIDE ETCHING OF W PROBE TIPS

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- Tungsten (W) probe tips are often used for wafer and die probing as well as for STM/AFM applications.
- Very sharp tips are needed – sharper than what can be achieved through standard mechanical machining processes.
- Concentrated (1-3 M) KOH or NaOH can be used to electrolytically etch W. A W wire is positively biased as the anode and the etch proceeds as:
  - Anode reaction:  $W(s) + 6OH^- \rightarrow WO_3(s) + 6e^- + 3H_2O$   
 $WO_3(s) + 2OH^- \rightarrow WO_4^{2-}(aq) + H_2O$
  - Cathode reaction:  $6H_2O + 6e^- \rightarrow 3H_2(g) + 6OH^-$
  - Typically, +4 to +12 Volts is applied to the W anode.
  - Higher voltages and higher concentrations produce faster etch rates and better tip results.

## HYDROXIDE ETCHING OF LARGE TIPS - SUCCESSIVE DIPS



The tip etches faster because the electric current density is concentrated there, and the tip is also in the solution longer.

## TECHNIQUES FOR REPRODUCIBLE WET ETCHES

- Design the etch rate for a controllable time period, e.g. several minutes, so that dip and extract times are comparatively short. Dilute the etch if necessary.
- Agitate gently using  $N_2$  bubblers, when available.
- Use a timer – don't guess.
- Use fresh chemicals.
- Control the temperature of the etch bath.
- Use a thermometer – don't guess.
- Never mix different substrates within the same etch bath.
- Always etch the same thing in the same equipment with the same chemicals under the same conditions!

## ETCH TUNING TECHNIQUES

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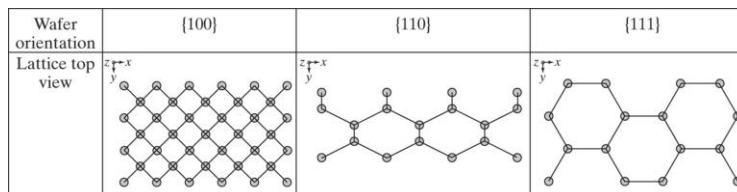
- For three-part etches (acid/base – oxidizer – diluent):
  - Adjust the overall etch rate through the amount of diluent while keeping the acid/base – oxidizer ratio constant.
  - Adjust the surface finish of the etch through the acid/base – oxidizer ratio while keeping the diluent fraction constant.
- If necessary, use two etches in sequence:
  - 1. Roughing etch with a high rate to rapidly remove material.
  - 2. Finishing etch with a slower etch rate and more polishing action to produce a higher quality final surface.
- Adjust the etch temperature to balance the reactant transport against the surface reaction rate.
  - Polishing etches tend to be transport-limited, usually by diffusion.
  - Roughing etches tend to be surface reaction rate-limited.

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## ANISOTROPIC WET ETCHING (11.3 & CHAPTER 20)

## SILICON CRYSTAL STRUCTURE

- Silicon atoms in a crystal lattice are regularly arranged in a lattice structure.
- Material properties, such as Young's modulus of elasticity, mobility, and piezoresistivity) and chemical etch rates of silicon bulk are orientation dependent.



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